

Environmental Management of Nutrients

Hunter Landis



VA DCR

 The Virginia Department of Conservation and Recreation (DCR) coordinates and directs programs and services to prevent degradation of the commonwealth's water quality caused by nonpoint source pollution.





Environmental Management of Nutrients <u>Knowledge Areas</u>:

- Factors causing decline of Chesapeake Bay
- Effects of nutrients in ground and surface waters
- Hydrologic cycle
- Nutrient loss mechanisms to ground and surface waters
- Identification and management of environmentally sensitive sites
- Seasonal nutrient loss patterns
- Use of cropping systems to reduce nutrient loss



Introduction

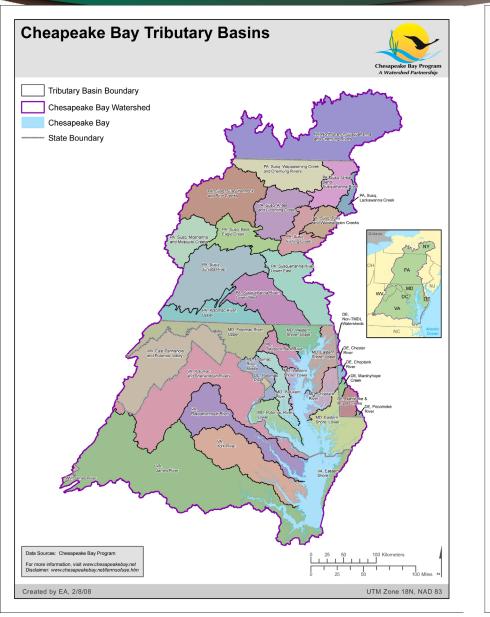
- Background: Education/Experience
- Involvement with Nutrient Management

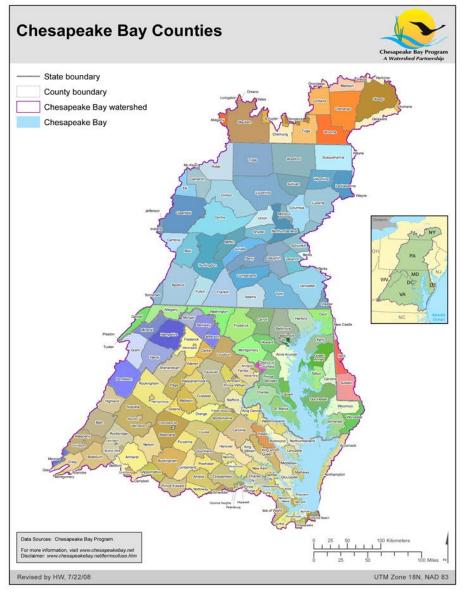


The Chesapeake Bay Watershed

- Is the largest estuary in the US and 3rd largest in the World.
- Supports more than 17 million people.
- Covers 64,000 square miles.
- Includes parts of Virginia, Maryland, Delaware, West Virginia, Pennsylvania, New York, and Washington, D.C.
- Has 11,684 miles of shoreline, more than the entire US West Coast.
- Contains more than 100,000 rivers and streams.
- Helps filter and protect the drinking water of 75% of Bay watershed residents.



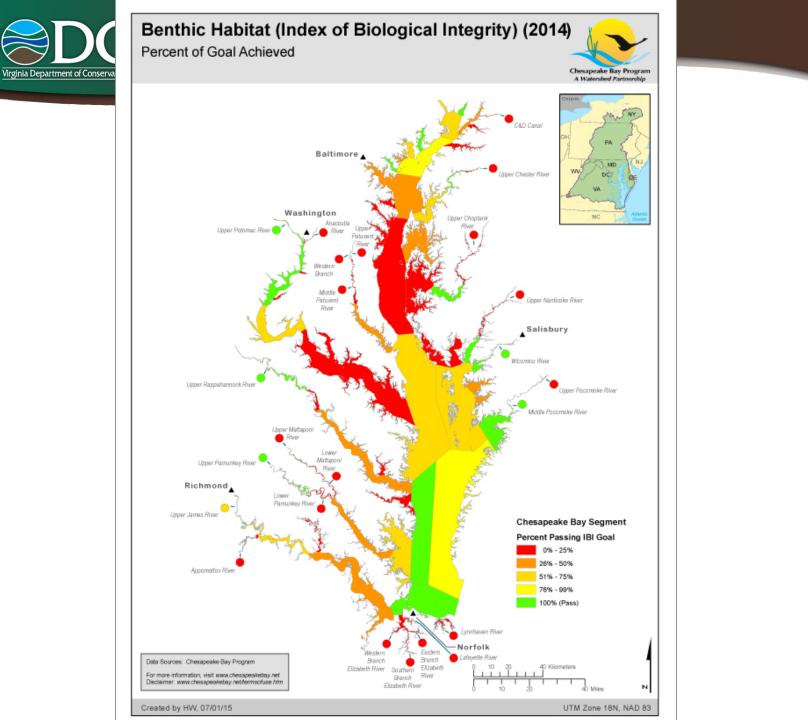






The Chesapeake Bay Watershed

- In 2010, the U.S. Environmental Protection Agency established the Chesapeake Bay Total Maximum Daily Load, a comprehensive cleanup plan to guide federal, state, local, and individual actions to reach the goal of a clean Chesapeake Bay and connecting waterways by 2025.
- In Virginia, the TMDL calls for:
 - 20.5% reduction in Nitrogen
 - 25.2% reduction in Phosphorous
 - 20.8% reduction in Sediment delivered to the bay
- Cleanup efforts will reduce flooding, protect groundwater, increase property value, restore fish and wildlife habitats, and improve air quality





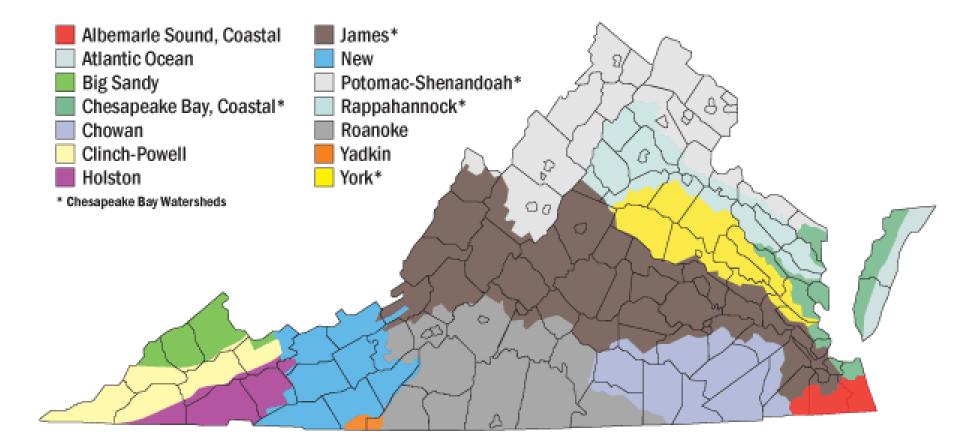
 CBF's health index, called the State of the Bay Report, estimates that the Chesapeake Bay watershed rated 100 on a scale of 100 in the 1600s. In 2018, the report rated the Bay at 33 out of 100. Water quality is so poor that the Chesapeake Bay is on the **Environmental Protection Agency's** "dirty waters" list. (https://www.cbf.org/issues/agriculture/nitrogenphosphorus.html)

Bay health stressed due to elevated temperatures

Virginia Department









Plans for Improved Water Quality

- Implementing nutrient management and conservation plans
- Planting cover crops





Plans for Improved Water Quality

- Fencing animals out of streams
- Installing and maintaining grassed or forested buffer strips along farm fields





Plans for Improved Water Quality

- Important natural filters such as forests, oysters, wetlands, and underwater grasses need to be protected and restored.
- Overall, the Bay has lost 98 percent of its oysters, about 80 percent of grasses, and nearly 50 percent of forest buffers.



Effects of nutrients in ground and surface waters

- Almost 40% of U.S. waters that have been assessed have not met water quality standards (Zygmunt, 2000).
- About 15,000 water bodies are impaired from siltation, nutrients, bacteria and other pathogens, oxygen-depleting constituents, trace elements, pesticides, and other organic chemicals (EPA).



Non-point source pollution

 Many of these pollutants do not come from a single point such as a sewage outfall or an industrial discharge pipe and are thus termed <u>non-point source</u> pollution.





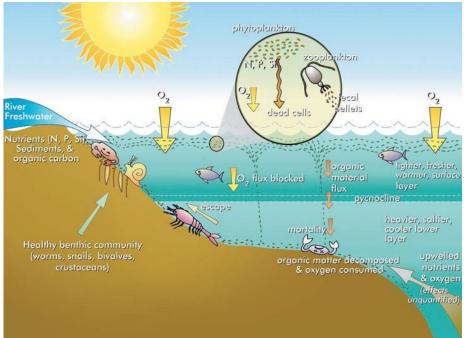
Non-point source pollution

- Nitrogen (N) and phosphorus (P), are the major pollutants in lakes and estuaries and the second leading source of pollution in rivers.
- Life within rivers, streams, lakes, and bays could not exist without nutrients; however, an <u>excess</u> of nutrients (*eutrophication*) may cause ecological problems and can harm aquatic life.



Eutrophication

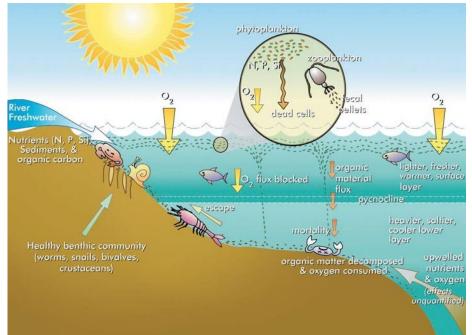
Eutrophication occurs when excessive nutrients cause a dense growth of algal blooms that can be seen. As algae ultimately die off and decompose, oxygen is consumed resulting in low levels of oxygen in the water.





Eutrophication

- Nutrients can come from many sources, including any of the following:
 - Fertilizers from <u>agriculture</u>, <u>golf courses</u>, and <u>suburban lawns</u>
 - <u>Erosion</u> of soil full of nutrients
 - Discharges from sewage treatment plants
 - Deposition of atmospheric nitrogen





Environmental Impact on Water Quality

- Nearly all of the N and P exported from watersheds in the Mid-Atlantic are from nonpoint sources, to which fertilizer and animal manures used in agriculture contribute significantly.
- A six-year study by the U.S. EPA (1983) revealed that runoff from farmland was a major source of pollution contributing to water quality decline in the Chesapeake Bay.



Environmental Impact on Water Quality

The largest source of pollution to the Bay comes from agricultural runoff, which contributes roughly
40 percent of the nitrogen and 50 percent of the phosphorus entering the Chesapeake Bay.



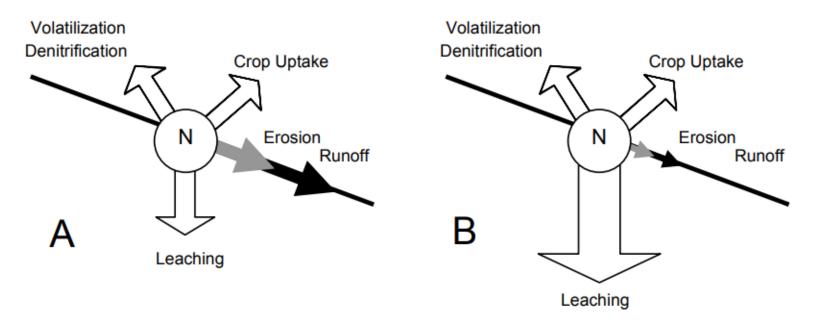


Table 1.1. Point and non-point source contributions to total nitrogen and phosphorus export from Mid-Atlantic watersheds.

			Non-point source			
Nutrient	Total export ^a	Point source	Fertilizer	Animal agriculture	Atmosphere	Non- agricultural runoff
	kg/ha/yr	Median, as % of total export				
Nitrogen	9.0	4	14	16	32	22
Phosphorus	0.68	14	19	25	NA	22
^a Total export is	the median	export fron	n hydrologic	cataloging.		



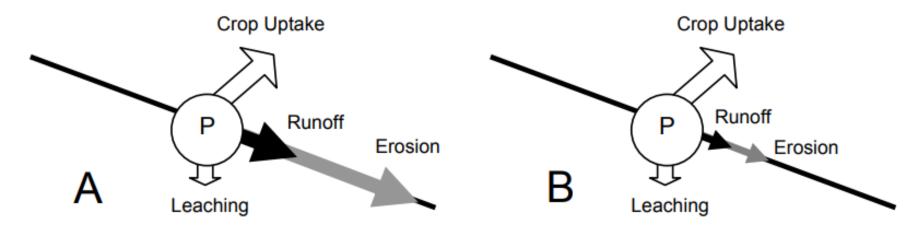
Figure 1.1. (A) General fate of N and (B) how adopting processes to reduce erosion and runoff increases N leaching losses.



 Managing N to minimize NO3 - losses is very difficult because of the many possible loss pathways. For example, increased water infiltration may increase leaching of nitrate if practices to reduce runoff and erosion, such as no-till, are adopted

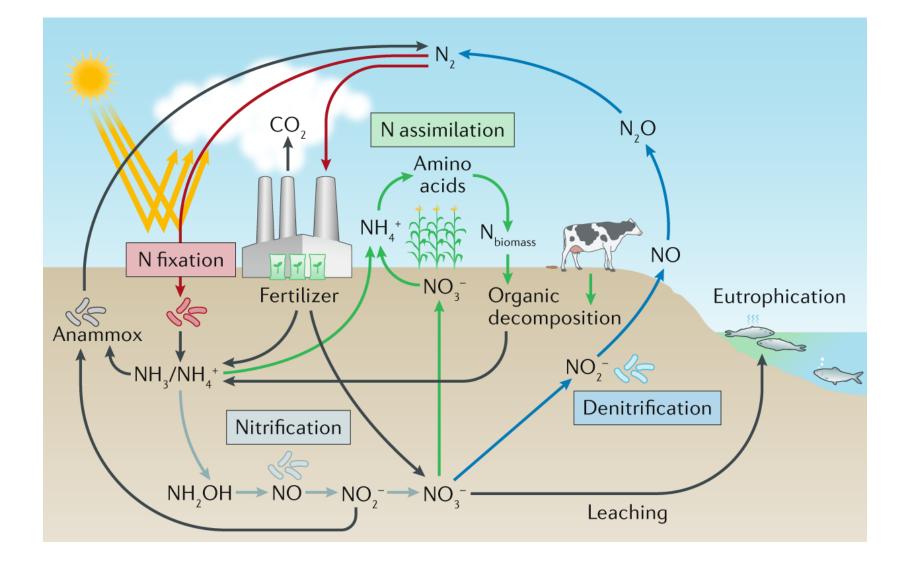


Figure 1.2. (A) General fate of P and (B) how adopting processes to reduce erosion and runoff does not usually increase P leaching losses.

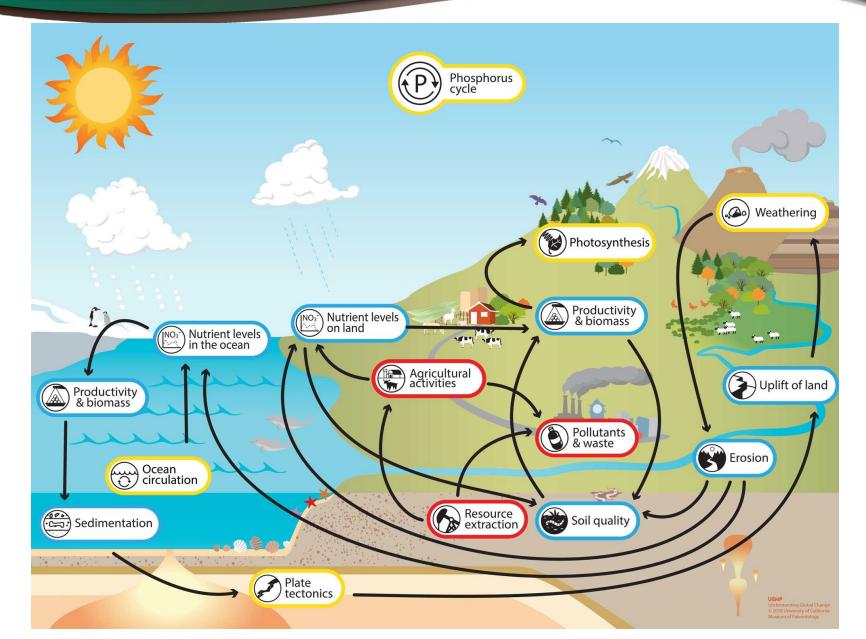


 Phosphorus is typically immobile in soil and seldom migrates downward with soil water to any great extent because it is strongly adsorbed by and/or precipitated as highly insoluble soil mineral phases.



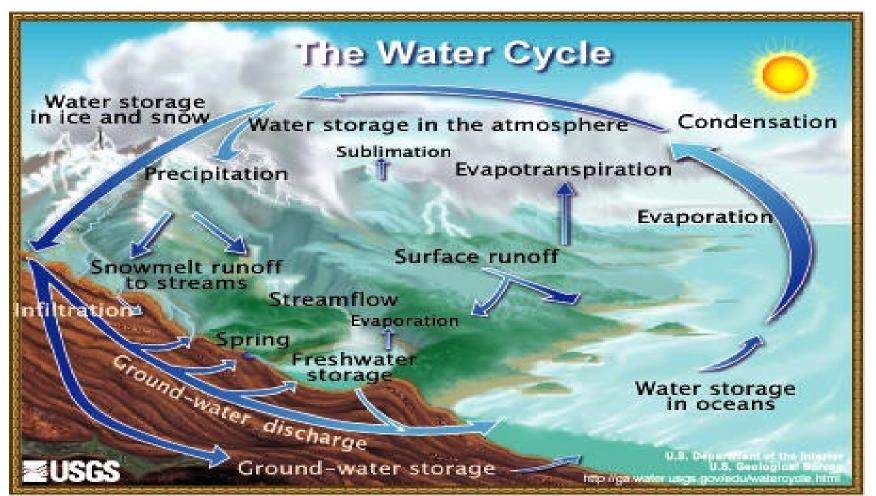








Hydrologic Cycle





Seasons of Greatest Leaching

 Leaching potential increases during times of low evapotranspiration and little plant growth & uptake

- Late fall
- Winter
- Early spring



Manure Spreading Schedule

CROP	JAN	FEB	MAR	APR	MAY	JUN	JUL	AUG	SEP	ОСТ	NOV	DE
ALFALFA												
BERMUDAGRASS												
CORN												
COTTON												
SMALL GRAIN												
SORGHUM												
SOYBEANS												
HAY **												
PASTURE**												
	** Ex	cept for <i>l</i>	Alfalfa, Bern	nuda grass	and othe	r warm s	eason gr	asses.				
			pread during									
	P	oultry lit	ter may be a	applied dur	ing these	times pr	ovided s	oil conditi	ions are			
			anure to from									



Nutrient Cycling on Farms

• Different types of farming operations have different ways to cycle nutrients.



Figure 1.3. Nutrient flows in modern animal agriculture.

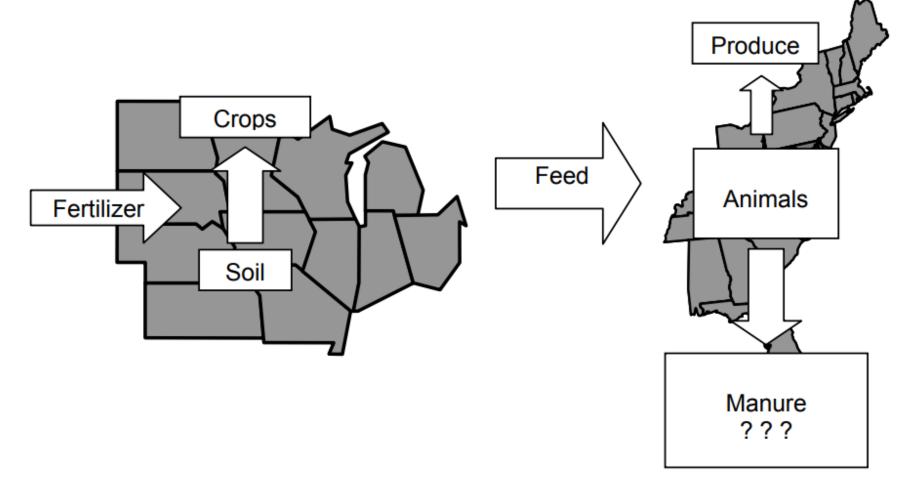




Figure 1.4. Nutrient cycles on cash crop farms.

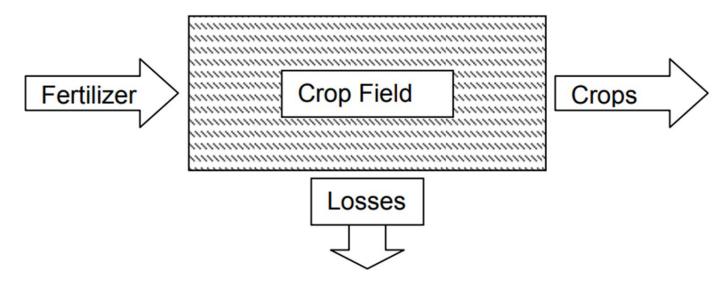


Table 1.2. Example of nutrient balance (P_2O_5) on a cash-crop farm in Pennsylvania.

Input:	lb P ₂ O ₅ /A/yr
Fertilizer	36
Output:	
Crop removal	32
Balance	+4



Figure 1.5. Nutrient cycles on a modern crop and livestock farm.

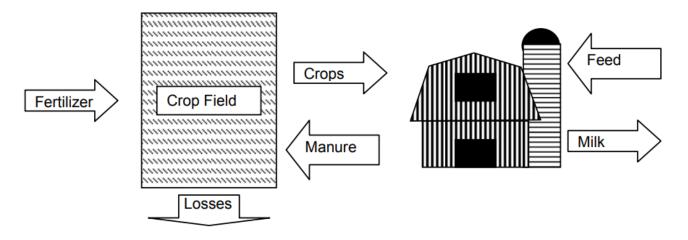


Table 1.3. Example of nutrient balance on a dairy farm in Pennsylvania.

Inputs:	lb P ₂ O ₅ /A/yr			
Fertilizer	22			
Feed	60			
Output:				
Milk	24			
Balance	+58			



Figure 1.6. Intensive animal production farm with limited crop production.

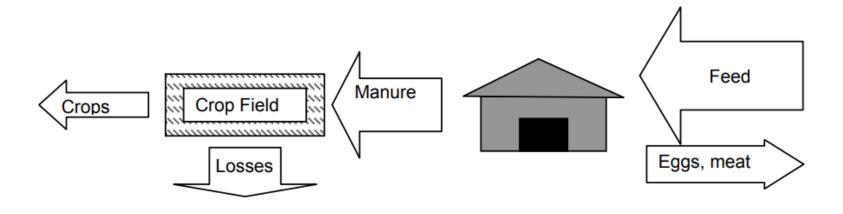


Table 1.4. Example of nutrient balance on a poultry layer farm in Pennsylvania.

Inputs:	lb P ₂ O ₅ /A/yr			
Fertilizer	0			
Feed	3380			
Output:				
Eggs	1030			
Balance	+2350			



Nutrient Impact in Surface Waters





Sedimentation

- Occurs when water carrying eroded soil particles slows long enough for soil particles to settle out.
- Effects water quality physically, chemically and biologically
- Destroys fish spawning beds, reduces useful storage volume in reservoirs, clogs streams, and make expensive filtration necessary for municipal water supplies.



Sediment

- Carries organic matter, animal or industrial wastes, nutrients, and chemicals.
- Most troublesome is phosphorous from fertilizers, organic matter and animal manure.
- May carry pesticides such as herbicides and insecticides that are toxic to plants & animals



Groundwater





Groundwater wells

- 21 percent of all Virginians have their own private water systems such as wells
- Homeowners are totally responsible for water quality testing
- The Virginia Household Water Quality Program through Virginia Cooperative Extension & BSE at VT provides water testing and education to private landowners
- Reach a minimum of 60 counties each year, conducting clinics and water sample kits



Household Water Quality Program

- Water samples are tested for 12 chemical constituents and for coliform bacteria
- In 2019, 64 such clinics were conducted covering 88 counties
- VT- BSE performed tests on 2294 wells in 2019



Water Quality Program Results

- 42% of all wells sampled did Not meet EPA Water Quality Standards for Coliform Bacteria
- 8% exceeded limits for Lead
- 71% of all participants stated that they had never tested their water
- Copper tested high in 18%



Nitrogen Groundwater Concerns

- Nitrate-nitrogen is mobile in the soil
- Can leach to groundwater
- Nitrate form most problematic
- 10.0 ppm nitrate + nitrite nitrogen EPA drinking water standard
- Consumption of high nitrate water by infants potentially dangerous
- "Blue Baby Syndrome" is a lack of oxygen transport to brain. There have been reported cases of Blue Baby Syndrome in Va.
- Evidence of livestock reproductive problems
- Drinking Water Nitrate Violations have doubled in the last 8 yrs.



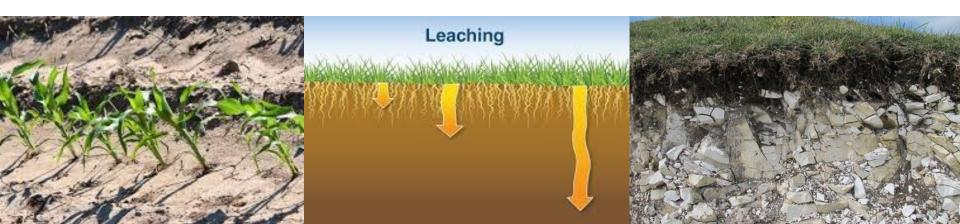
Runoff and Leaching

- Dissolved nutrients and pesticides can reach groundwater by moving down through the soil. Nitrogen moves this way.
- Certain pesticides are highly mobile and have been detected in groundwater.
 Aldicarb (Temik), alachlor (Lasso), and triazines (Atrazine) are just a few.



Environmentally Sensitive Site

- Environmentally sensitive site" means any field which is particularly susceptible to nutrient loss to groundwater or surface water since it contains, or drains to areas which contain, *sinkholes, or where at least 33% of the area in a specific field contains one or any combination* of the following features:
- 1. Soils with <u>high potential for leaching</u> based on soil texture or excessive drainage;
- 2. <u>Shallow soils</u> less than 41 inches deep likely to be located over fractured rock or limestone bedrock;





Environmentally Sensitive Site

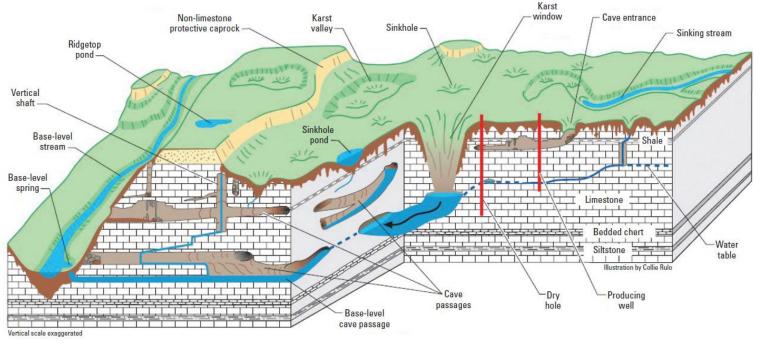
- 3. Subsurface tile drains;
- 4. Soils with high potential for subsurface lateral flow based on soil texture and/or **poor drainage**;
- 5. Floodplains as identified by soils **prone to frequent flooding** in county soil surveys; or
- 6. Lands with slopes greater than 15%.







Karst Topography



- Underlying limestone formations which may be characterized by solution cavities or "<u>sinkholes</u>" which form a *direct connection between surface* and groundwater due to collapse of the soil profile into the cavity.
- Pollution sources can be some distance away



Karst Topography where multiple issues occur





Determining Environmentally Sensitive Sites

Use site visit and soil survey - Do areas of the field have one or more sinkholes or does part of the field drain to a sinkhole?

- Or does at least 33% of the field have any combination of the following:
 - From Table 1-4 Standards and Criteria pages 28-36
 - soils with a "H" for environmental sensitivity
 - a. Leaching
 - b. Shallow soils
 - c. Drainage Soils with high potential for subsurface lateral flow

(continued on next slide)



Determining Environmentally Sensitive Sites -Continued

From site visit –

- d. Subsurface tile drains
- e. Soils with very slow permeability rates/high run off potential
- From soil survey –
- f. Floodplains soils prone to "frequent" flooding (usually in soil and water features table)
- g. Lands with slopes greater than 15%
- "E" slope or greater in Coastal Plain
- "D" slope or greater in other regions



Nitrogen Loss Forms & Pathways

- NH₄⁺ bound to eroding sediment or organic matter
- Organic N suspended in runoff water
- Soluble NO₃⁻ in runoff water
- NO₃⁻ leaching to groundwater



Degree of Nitrate Leaching

Precipitation amounts and timing

Physical properties of soil

• Nitrate levels in soil



Nutrient Practices to Reduce Nitrogen Pollution Potential

- Rate of application
- Timing of application
- Placement of nutrients
- Cover crops (Trap crops)



Timing of Applications

• When is the best time to apply nutrients to crop?

• When it needs it Most!



Phosphorus Loss Forms & Pathways

- Particulate P complexes eroded from soil with sediment. The smaller the particle, the longer it stays in suspension.
- Organic P suspended in runoff water
- Soluble HPO_4^{-2} or $H_2PO_4^{-1}$ in runoff water
- Soluble P in subsurface flow and tile drains (mainly course textured poorly drained soils)



Nutrient Practices to Reduce Phosphorus Pollution Potential

- Keep Soil Surface P Saturation Levels Below Environmentally Critical Levels
- Reduce Soil Erosion on Land With Extreme Levels of Soil Test P and on Highly Erodible or Highly Leachable Land
- Keep P Applications Below Crop Removal Rates in High Risk Situations



Nitrogen vs Phosphorous Management Strategies

- Nitrogen
 - Rate-based upon Crop Needs
 - Timing- when plants most need
 - Placement- in root zone or banding
 - Cover crops- scavenge residual N from previous crop





Nitrogen vs Phosphorous Management Strategies

- Phosphorous
 - Erosion Control- particulate P- Target
 - Manage runoff -organic P + Plant Avail P
 - Contour Farming Terraces
 - Concentrations of soil test P Source
 - Reduce P applications incorporate to reduce P concentrations

