



# **Commercial Fertilizer**



#### References

 Mid-Atlantic Nutrient Management Handbook

– Chapter 8: Commercial Fertilizers



# **Commercial Fertilizer**

- Nitrogen
- Phosphorous
- Potassium
- (Sulfur)
- (Lime)



#### Nitrogen

 Inorganic N fertilizers are produced by fixing N from the atmosphere

- Natural gas as the energy source



# **Nitrogen Fertilizers**

- Urea [CO(NH<sub>2</sub>)<sub>2</sub>]:
  - Fertilizer grade: 46-0-0
  - Soluble, readily available source of N
  - Dry fertilizer product
  - Produced by reacting ammonia (NH<sub>3</sub>) with carbon dioxide under pressure at an elevated temperature
  - Contains the highest percentage of N of all dry fertilizers
- Ammonium nitrate  $(NH_4NO_3)$ :
  - Fertilizer grade: 34-0-0
  - Soluble, readily available source of N; non volatile and non leeching
  - Dry fertilizer product
- Ammonium sulfate [(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>]:
  - Fertilizer grade: 21-0-0-24S
  - Contains 24% sulfur
  - Soluble, readily available source of N and S
- Non-pressure nitrogen solutions:
  - Fertilizer grade: ranges from 28-0-0 to 32-0-0
  - Soluble, readily available source of N
  - Liquid fertilizer product that does not require pressure for storage
  - Usually referred to as UAN (Urea Ammonium Nitrate)
- Aqua ammonia ( $NH_4OH$ ):
  - Fertilizer grade: 20-0-0 (most common)
  - Density of 20-0-0 is 7.60 lbs/gal at 60°F
  - Produced by dissolving NH<sub>3</sub> gas in water



#### Phosphorous

- Ingredient for producing phosphorus (P) fertilizers is rock phosphate
  - Comes from the mineral apatite, a calcium
- Most conventional P fertilizers are made by reacting rock phosphate with:
  - Sulfuric acid to create superphosphate (20%)
  - Phosphoric acid to create triple superphosopate (46%)



### **Phosphorus Fertilizer**

- Diammonium phosphate [(NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>]:
  - Fertilizer grade: 18-46-0
  - Soluble, readily available source of P and N
  - Dry fertilizer product
- Monoammonium phosphate (NH<sub>4</sub>H<sub>2</sub>PO<sub>4</sub>):
  - Fertilizer grade: 11-52-0
  - Soluble, readily available source of P and N
- Ammonium polyphosphate [(NH<sub>4</sub>)n+2PnO<sub>3</sub>n+1]:
  - Fertilizer grade: 10-34-0 or 11-37-0
  - Soluble, readily available source of P and N
  - Liquid fertilizer product
- Concentrated (Triple) superphosphate [Ca(H<sub>2</sub>PO<sub>4</sub>)2•H<sub>2</sub>O]:
  - Fertilizer grade: 0-46-0
  - Soluble, readily available source of P
  - Dry fertilizer product



#### Potassium

- Term "potash"
  - Comes from an early production technique where potassium was leached from wood ashes and concentrated by evaporating the leachate in large iron pots ("pot-ash")
- Potash is a potassium-rich salt that is mined from underground deposits formed from evaporated sea beds millions of years ago
- \*\*Elemental potassium is K
- \*\*Potash is K<sub>2</sub>O



### **Potassium Fertilizer**

- Potassium chloride (Muriate of Potash) (KCI):
  - Most abundantly used form of potassium fertilizer
  - Contains 60-63% K<sub>2</sub>O
  - Water soluble source of K
- Potassium sulfate  $(K_2SO_4)$ :
  - Contains 50-53%  $K_2O$ , 18% S, and no more than 2.5% Cl
- Potassium-magnesium sulfate (sul-po-mag) (K<sub>2</sub>SO<sub>4</sub>•2MgSO<sub>4</sub>):
  - Contains about 22% K<sub>2</sub>O, 11% Mg, 22% S, and no more than 2.5% Cl
  - Along with the K, this product is a good source of Mg and S
- Potassium nitrate (KNO<sub>3</sub>):
  - Contains about 44% K2O and 13% N



# **Fertilizer Analysis**

- Expressed as **PERCENTAGES** 
  - -N = % nitrogen
  - -P = % phosphorus
  - K = % potassium (potash)
  - -S = % Sulphur
- N-P-K-S



## **Fertilizer Materials**

- 4 Categories, 2 Types
  - Granular
    - Dry Bulk Blend
    - Ammoniated Dry Granular
  - Liquid
    - Clear Liquid Solution
    - Liquid Suspension



# **Dry Bulk Blends**

- A blend is made by mixing two or more fertilizer materials
  - Particles of nitrogen, phosphate, potash, limestone filler, and some secondary nutrients and micronutrients Produced by blending various dry materials
- Custom blending to meet exact nutrient needs of the crop



# **Dry Bulk Blends**

- Important to match particle size and density
- Reduces segregation during transport to the field
- Larger particle and less dense materials tend to move to the top of the load
- Similar particles improve uniformity during application

- more dense particles throw farther



# **Dry Bulk Blends**

- Common dry nitrogen (N) materials – ammonium sulfate 21-0-0-24
  - UREA 46-0-0
- Common Dry Phosphate (P) materials
  - Diammonium phosphate (DAP) 18-46-0
  - concentrated (triple) superphosphate 0-46-0
- Common Dry Potassium (K) materials
  muriatic of potash (potash) 0-0-60
  - potassium-magnesium sulfate 22-22-11
    - (sul-po-mag or k-mag)



















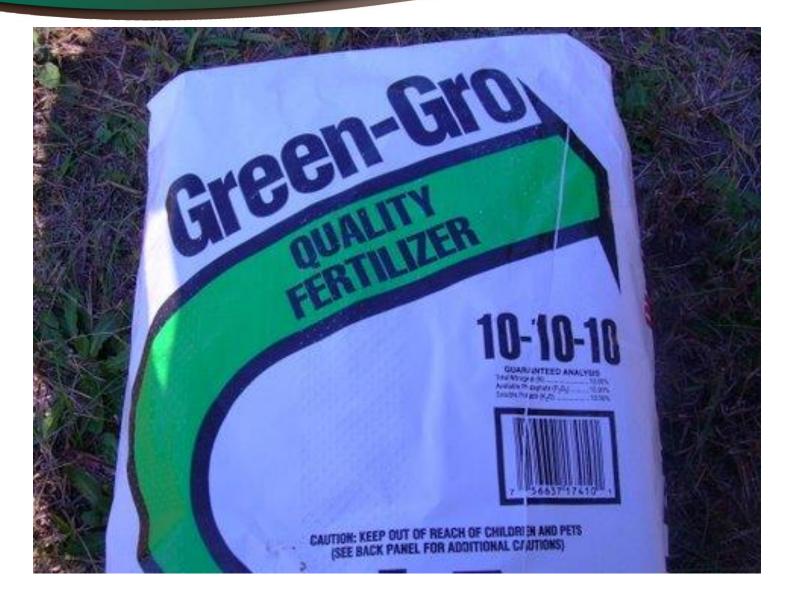
# **Ammoniated Dry Granular**

- In ammoniated fertilizers materials supplying N, P, K, etc.. are chemically combined into one solid granule
  - The granules' size, shape and density are closely controlled so that each granule has consistent percentages of each nutrient.
- Ammoniated fertilizer distribution breakdown is not possible because the nutrients are chemically combined, resulting in a consistent application











## **Bulk Blended vs Ammoniated**

- Ammoniated
  - Chemically manufactured for homogenous blended of N-P-K in each granule
  - Better for lower analysis (banding along rows)
- Bulk Blends
  - Mixed together physically; segregate during transportation, handling and application, resulting in uneven distribution in the field
  - Can be mixed on site to meet specific recommendations
  - Higher analysis
  - Less manufacturing costs



# **Clear Liquids**

- Cold mix process
  - Start 10-34-0 base
  - commonly used as a starter
  - additional nitrogen or potash sources may be added for a complete fertilizer



# **Clear Liquids**

- Hot mix process
  - aqua ammonia reacted with phosphoric acid
    - (generates considerable heat)
  - water (filler), potash, and additional nitrogen added to mix



# **Clear Liquids**

- Hot mix process
  - common liquid materials
    - aqua ammonia 20-0-0
    - phosphoric acid
    - water
    - potash 0-0-60
    - 28%-32% UAN nitrogen solution
    - micronutrients





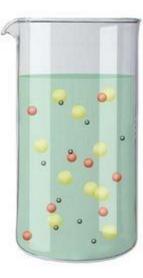


# **Liquid Suspension**

- High analysis with clear liquids is limited
  - Clay provides a surface for the fertilizer salts to form on
- Nutrients are held in "suspension"
  - Suspensions allow for less material per acre (higher analysis)
- Flexibility
  - Allows for more acres per load
- Agitation necessary to prevent settling of clay



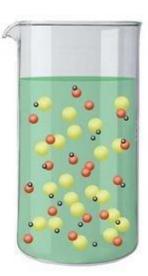
#### **Clear Solution**



Macro Nutrients



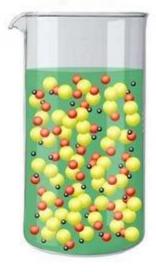
#### Saturated Solution



Sub Macro Nutrients



#### Water Soluble Suspension



**Micro Nutrients** 

- •Fe •Mn •Zn •Cu
  - 211 Cu
- B Co
- Mo







#### **Advantages of Granular and Liquid**

- Liquid
  - Ease of handling and application (once set up)
  - Ease of blending
  - Uniformity of application
  - Starter and in-season application
  - Blend with crop protection products
- Granular
  - Cheaper in bulk
  - Easier to store (does not "settle out" over time or "salt out" in cold weather)
  - More efficient for heavy pre-plant applications
  - Slow-release options (polymer-coated urea)



### Disadvantages

- Spatial:
  - distance from plant roots to fertilizer nutrients
  - less mobile nutrients (phosphorus) can't get closer than the individual granule containing them
  - in liquid form more mobile in the soil water solution
- Salt content
  - granular fertilizers can be "hot"
  - roots can steer away from a band of granular fertilizer that contains high levels of nitrogen and potassium
- Consistency
  - nutrient content is identical in every drop of liquid fertilizer
  - granular have individual nutrient components in each granule
- Equipment
  - cost of converting equipment to handle liquid fertilizer can be \$\$



#### **TerraGator**











#### **RoGator**











#### **Nurse Trucks**











#### **Tender Trucks and Trailers**











## **Dry Truck**









#### **Sidedress**

